

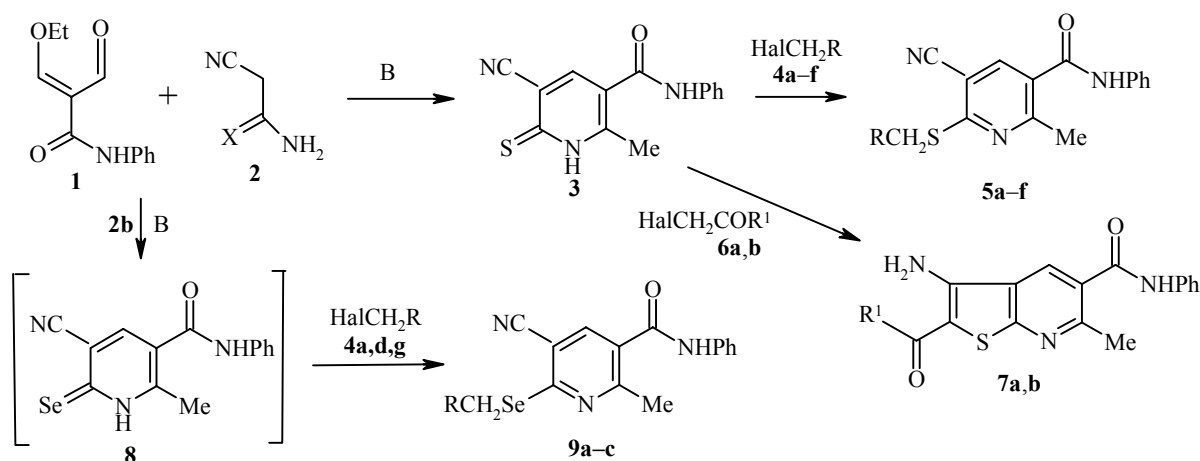
SYNTHESIS OF DERIVATIVES OF 3-CYANO-6-METHYL-5-PHENYLCARBAMOYLPYRIDINE-2(1H)-THIONE AND 3-CYANO-6-METHYL-5-PHENYLCARBAMOYLPYRIDINE-2(1H)-SELENONE

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The reaction of ethoxymethylenacetanilide with cyanothioacetamide or cyanoselenoacetamide in the presence of *N*-methylmorpholine and alkylating agents gave substituted 2-alkylthio- and 2-alkylselenopyridines and thieno[2,3-*b*]pyridines.

Keywords: 2-alkylthiopyridines, 2-alkylselenopyridines, thieno[2,3-*b*]pyridines.

4-R-3-Cyano-6-methyl-5-phenylcarbamoilpyridine-2(1H)-thiones have been synthesized through the condensation of aldehydes, the anilide of acetoacetic acid, and cyanothioacetamide. Depending on the nature of the aldehyde used, 4-alkyl- [1, 2], 4-aryl- [3,4], 4-hetaryl- [5-8], or 4-cyclohexanespiropyridinethiones [9] were obtained. Interest in these pyridinethione derivatives is related to their liver protection activity [10]. Since the 4-unsubstituted analogs of these compounds have not been reported, we developed a method for the synthesis of derivatives of 3-cyano-6-methyl-5-phenylcarbamoilpyridine-2(1H)-thiones and 3-cyano-6-methyl-5-phenylcarbamoilpyridine-2(1H)-selenones.



2 a X = S, b X = Se; 4, 5 a R = H, Hal = I; b R = Et, Hal = Br; c R = H₂C=CH, Hal = Br; d R = Ph, Hal = Cl; e R = H₂NC(O), Hal = I; f R = 3-cyano-6-methyl-2-methylthiopyridin-5-yl, Hal = Br; g R = Me, Hal = I; 6, 7 a R¹ = 4-BrC₆H₄, Hal = Br, b R¹ = 3-cyano-6-methyl-2-methylthiopyridin-5-yl, Hal = Br; 9 a R = H, b R = Me, c R = Ph; B = *N*-methylmorpholine

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TABLE 1. Characteristics of Compounds **3**, **5a-f**, **7a,b**, and **9a-c**

Compound	Empirical formula	Found, %				mp, °C*	IR spectrum, cm ⁻¹		¹ H NMR spectrum, δ, ppm, <i>J</i> (Hz)	Yield, %
		Calculated, %					C=N NH ₂	C=O NH		
1	2	3	4	5	6	7	8	9	10	11
3	C ₁₄ H ₁₁ N ₃ OS	<u>62.54</u> 62.43	<u>4.22</u> 4.13	<u>15.60</u> 15.59	<u>11.91</u> 11.92	250-252	2220	1640, 3200, 3320	2.53 (3H, s, CH ₃); 7.34-7.64 (5H, m, Ph); 8.30 (1H, s, C(4)H); 10.30 (1H, br. s, NHCO); 14.27 (1H, br. s, NH)	86
5a	C ₁₅ H ₁₃ N ₃ OS	<u>63.68</u> 63.56	<u>4.52</u> 4.63	<u>14.73</u> 14.84	<u>11.32</u> 11.32	190-192	2220	1640, 3270	2.66 (6H, br. s, CH ₃ and SCH ₃); 7.22-7.72 (5H, m, Ph); 8.38 (1H, s, C(4)H)	80
5b	C ₁₇ H ₁₇ N ₃ OS	<u>65.67</u> 65.58	<u>5.50</u> 5.58	<u>13.49</u> 13.49	<u>10.40</u> 10.31	160-162	2220	1640, 3250	1.00 (3H, t, <i>J</i> = 6.4, CH ₃ CH ₂); 1.71 (2H, m, CH ₃ CH ₂); 3.31 (2H, t, <i>J</i> = 5.9, SCH ₂); 7.10-7.72 (5H, m, Ph); 8.36 (1H, s, C(4)H); 10.45 (1H, br. s, NH)	92
5c	C ₁₇ H ₁₅ N ₃ OS	<u>66.10</u> 66.02	<u>4.99</u> 4.88	<u>13.48</u> 13.58	<u>10.39</u> 10.35	157-159	2210	1640, 3290	2.65 (3H, s, CH ₃); 3.99 (2H, d, <i>J</i> = 6.4, SCH ₂); 5.13 (d, <i>J</i> = 9.1) and 5.34 (d, <i>J</i> = 15.9), (for 1H, CH ₂ =); 5.93 (1H, m, CH=); 7.08-7.70 (5H, m, Ph); 8.36 (1H, s, C(4)H); 10.42 (1H, br. s, NH)	88
5d	C ₂₁ H ₁₇ N ₃ OS	<u>70.17</u> 70.15	<u>4.67</u> 4.76	<u>11.79</u> 11.70	<u>8.82</u> 8.94	161-163	2220	1640, 3250	2.69 (3H, s, CH ₃); 4.58 (2H, s, SCH ₂); 7.20-7.70 (10H, m, CH ₂ Ph and HNPh); 8.37 (1H, s, C(4)H); 10.46 (1H, br. s, NH)	85
5e	C ₁₆ H ₁₄ N ₄ O ₂ S	<u>58.88</u> 58.87	<u>4.42</u> 4.30	<u>17.27</u> 17.19	<u>9.72</u> 9.83	208-210	2220, 3400	1640, 1660, 3290	2.71 (3H, s, CH ₃); 4.02 (2H, s, SCH ₂); 6.98 br. s and 7.467 br. s (for 1H, NH ₂); 7.07-7.79 (5H, m, Ph); 8.23 (1H, s, C(4)H); 10.30 (1H, br. s, NH)	87

TABLE 1 (continued)

1	2	3	4	5	6	7	8	9	10	11
5f	C ₂₄ H ₁₉ N ₅ O ₂ S ₂	$\frac{60.97}{60.86}$	$\frac{4.04}{4.05}$	$\frac{14.89}{14.78}$	$\frac{13.64}{13.55}$	218 subl.	2225	1650, 1710, 3270	2.67 (6H, s, SCH ₃ and 6-CH ₃ methylthiopyridin-5-yl); 4.74 (2H, s, SCH ₂); 7.07-7.79 (5H, m, Ph); 8.25 s and 8.86 s (for 1H, C(4)H pyridine); 10.29 (1H, br. s, NH)	60
7a	C ₂₂ H ₁₆ BrN ₃ O ₂ S	$\frac{56.56}{56.65}$	$\frac{3.66}{3.47}$	$\frac{9.01}{9.00}$	$\frac{6.78}{6.89}$	302-305	3400	1650, 1660, 3270	2.27 (3H, s, CH ₃); 7.09-7.92 (9H, m, Ph and C ₆ H ₄); 8.43 (2H, br. s, NH ₂); 8.80 (1H, s, C(4)H); 10.45 (1H, br. s, NH)	89
7b	C ₂₄ H ₁₉ N ₅ O ₂ S ₂	$\frac{60.77}{60.85}$	$\frac{4.04}{4.05}$	$\frac{14.65}{14.76}$	$\frac{13.44}{13.54}$	279-281	2210, 3400	1650, 3270	2.59 s, 2.61 s and 2.69 s (for 3H, CH ₃); 7.07-7.74 (5H, m, Ph); 8.14 br. s and 8.81 br. s (for 1H, C(4)H pyridine); 8.45 (2H, br. s, NH ₂); 10.44 (1H, br. s, NH)	78
9a	C ₁₅ H ₁₃ N ₃ OSe	$\frac{54.45}{54.56}$	$\frac{3.87}{3.96}$	$\frac{12.72}{12.73}$	$\frac{23.81}{23.90}$	202-204	2220	1640, 3290	2.57 (3H, s, SeCH ₃); 2.66 (3H, s, CH ₃); 7.09-7.72 (5H, m, Ph); 8.33 (1H, s, C(4)H); 10.43 (1H, br. s, NH)	83
9b	C ₁₆ H ₁₅ N ₃ OSe	$\frac{55.92}{55.80}$	$\frac{4.39}{4.40}$	$\frac{12.11}{12.23}$	$\frac{22.84}{22.93}$	135-137	2230	1660, 3290	1.50 (3H, t, <i>J</i> = 6.9, CH ₃ CH ₂); 2.66 (3H, s, CH ₃); 3.30 (2H, s, CH ₃ CH ₂); 7.12-7.72 (5H, m, Ph); 8.33 (1H, s, C(4)H); 10.43 (1H, br. s, NH)	75
9c	C ₂₁ H ₁₇ N ₃ OSe	$\frac{62.07}{62.06}$	$\frac{4.12}{4.23}$	$\frac{11.34}{11.33}$	$\frac{19.53}{19.44}$	184-186	2220	1680, 3300	2.73 (3H, s, CH ₃); 4.62 (2H, s, CH ₂); 7.02-7.79 (10H, m, CH ₂ Ph and HNPh); 8.31 (1H, s, C(4)H); 10.42 (1H, br. s, NH)	82

* EtOH for crystallization of **3**, DMF for other products.

The reaction of ethoxymethylenacetanilide (**1**) with cyanothioacetamide (**2a**) in the presence of N-methylmorpholine in absolute ethanol at 25°C gave 3-cyano-6-methyl-5-phenylcarbamoylpyridine-2(1H)-thione (**3**) as indicated by the spectral data. The IR spectra of thione **3** has bands for stretching vibrations of a conjugated nitrile group in the vicinity of 2220 cm⁻¹, imino groups at 3200-3320 cm⁻¹, and carbonyl group in the vicinity of 1640 cm⁻¹. The ¹H NMR spectra show characteristic signals for protons of the methyl group at 2.54 ppm, phenyl group and pyridine C(4)H at 7.34-7.64 and 8.30 ppm, and NH protons of the carbamoyl fragment and pyridine ring at 10.30 and 14.27 ppm, respectively.

Alkylation of thione **3** by halides **4a-f** in DMF in the presence of an equimolar amount of KOH proceeds regioselectively, leading to the corresponding alkylthiopyridines **5a-f**. The use of aroylmethyl halides (**6a**) and hetaroylmethyl halides (**6b**) in the alkylation of thione **3** in the presence of a two-fold excess of KOH leads to 3-amino-2-aroyle- (**7a**) and 3-amino-2-hetaroyl-6-methyl-5-phenylcarbamoylthieno[2,3-*b*]pyridines (**7b**).

The reaction of **1** with cyanoselenoacetamide **2b** in the presence of base in absolute ethanol in an argon atmosphere did not yield 3-cyano-6-methyl-5-phenylcarbamoylpyridine-2(1H)-selenone (**8**) as a pure compound, probably due to the strong tendency of selenone **8** to undergo oxidation [11]. However, 2-alkylselenopyridines **9a-c** were obtained upon introducing alkyl halides **4a,d,f** into the reaction mixture. The structure of these products suggests the intermediate formation of selenone **8**.

The structures of **3**, **5a-f**, **7a,b**, and **9a-c** are in accord with their physico-chemical characteristics (Table 1).

EXPERIMENTAL

The IR spectra were taken on an IKS-29 spectrometer for vaseline mulls. The ¹H NMR spectra were registered on a Bruker WP-100SY spectrometer at 100 MHz, Bruker AM-300 spectrometer for **5e** at 300.13 MHz, and Gemini-200 spectrometer for **5f**, **7a,b** at 199.975 MHz for solutions in DMSO-d₆ with TMS as the internal standard. The reaction course and purity of the products were monitored by thin-layer chromatography on Silufol-254 plates with 3:5 acetone–heptane as the eluent and iodine vapor as the developer.

3-Cyano-6-methyl-5-phenylcarbamoylpyridine-2(1H)-thione (3). A mixture of ethoxyethylene **1** (10 g, 42.9 mmol), cyanothioacetamide **2a** (4.29 g, 42.9 mmol), and N-methylmorpholine (4.8 ml, 42.9 mmol) was stirred in absolute ethanol at 25°C for 2 h. The precipitate formed was filtered off and washed with absolute ethanol and hexane to give thione **3** (Table 1).

2-Alkylthio-3-cyano-6-methyl-5-phenylcarbamoylpyridines (5a-f). A sample of 10% aqueous KOH (2 ml, 3.7 mmol) was added with stirring to a suspension of thione **3** (1 g, 3.7 mmol) in DMF (10 ml). After 5 min, the corresponding alkyl halide **4a-f** (3.7 mmol) was added, and the mixture was stirred for 4 h. The precipitate formed was filtered off and washed with ethanol and hexane to give **5a-f** (Table 1)

3-Amino-2-aroyle- (7a) and 3-amino-2-hetaroyl-6-methyl-5-phenylcarbamoylthieno[2,3-*b*]pyridines (7b). A sample of 10% aqueous KOH (2 ml, 3.7 mmol) was added with stirring to a suspension of thione **3** (1 g, 3.7 mmol) in DMF (10 ml). After 5 min, the corresponding aroyl halide (**6a**) or hetaroyl halide (**6b**) (3.7 mmol) was added. After 0.5 h, another 10% aqueous KOH (2 ml) was added and the mixture was stirred for an additional 4 h. The precipitate formed was filtered off and washed with ethanol and hexane (Table 1).

4-Alkylseleno-3-cyano-6-methyl-5-phenylcarbamoylpyridines (9a-c). A mixture of ethoxyethylene **1** (2 g, 8.57 mmol), cyanoselenoacetamide **2b** (1.3 g, 8.57 mmol), and N-methylmorpholine (0.96 ml, 8.57 mmol) in absolute ethanol was stirred for 2 h in an argon atmosphere at 25°C. Then, the corresponding alkyl halide **4a,d,f** (8.57 mmol) was added and the mixture was stirred for 4 h. The precipitate formed was separated and washed with ethanol and hexane to give **9a-c** (Table 1).

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REFERENCES

1. V. D. Dyachenko, S. G. Krivokolysko, and V. P. Litvinov, *Khim. Geterotsikl. Soedin.*, 1232 (1996).
2. V. D. Dyachenko, S. G. Krivokolysko, and V. P. Litvinov, *Khim. Org. Khim.*, **34**, 750 (1998).
3. V. D. Dyachenko, S. G. Krivokolysko, and V. P. Litvinov, *Khim. Geterotsikl. Soedin.*, 560 (1997).
4. V. D. Dyachenko, S. G. Krivokolysko, V. N. Nestorov, and V. P. Litvinov, *Khim. Geterotsikl. Soedin.*, 1243 (1996).
5. S. G. Krivokolysko, V. D. Dyachenko, and V. P. Litvinov, *Khim. Geterotsikl. Soedin.*, 672 (1997).
6. V. D. Dyachenko, S. G. Krivokolysko, and V. P. Litvinov, *Khim. Org. Khim.*, **34**, 927 (1998).
7. V. D. Dyachenko, S. G. Krivokolysko, and V. P. Litvinov, *Khim. Geterotsikl. Soedin.*, 666 (1997).
8. S. G. Krivokolysko, V. D. Dyachenko, and V. P. Litvinov, *Khim. Org. Khim.*, **33**, 1088 (1997).
9. V. D. Dyachenko, S. G. Krivokolysko, and V. P. Litvinov, *Khim. Geterotsikl. Soedin.*, 1533 (1997).
10. V. D. Krauze, A. G. Odynets, A. A. Verreva, S. K. Germane, A. N. Kozhukhov, and G. Ya. Dubur, *Khim.-Farm. Zh.*, **25**, No. 7, 40 (1991).
11. V. D. Dyachenko, *Chemical Sciences Candidate's Dissertation*, Moscow (1990).